

ASHIMYCINS A AND B, NEW STREPTOMYCIN ANALOGUES

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Detailed analysis of the fermentation broth of *Streptomyces griseus* strain FT3-4 resulted in the identification of two new streptomycin analogues named ashimycins A and B. Their structures have been determined by NMR spectral analysis and chemical degradations.

During the attempts to improve the fermentation yield of streptomycin by *Streptomyces griseus* strain FT3-4, we noticed that the lower production yield of streptomycin was sometimes accompanied by the accumulation of new metabolites structurally related to streptomycin. We will report herein the fermentation, isolation, physico-chemical properties, structural elucidation and biological activities of these compounds named ashimycins A and B.

Fermentation

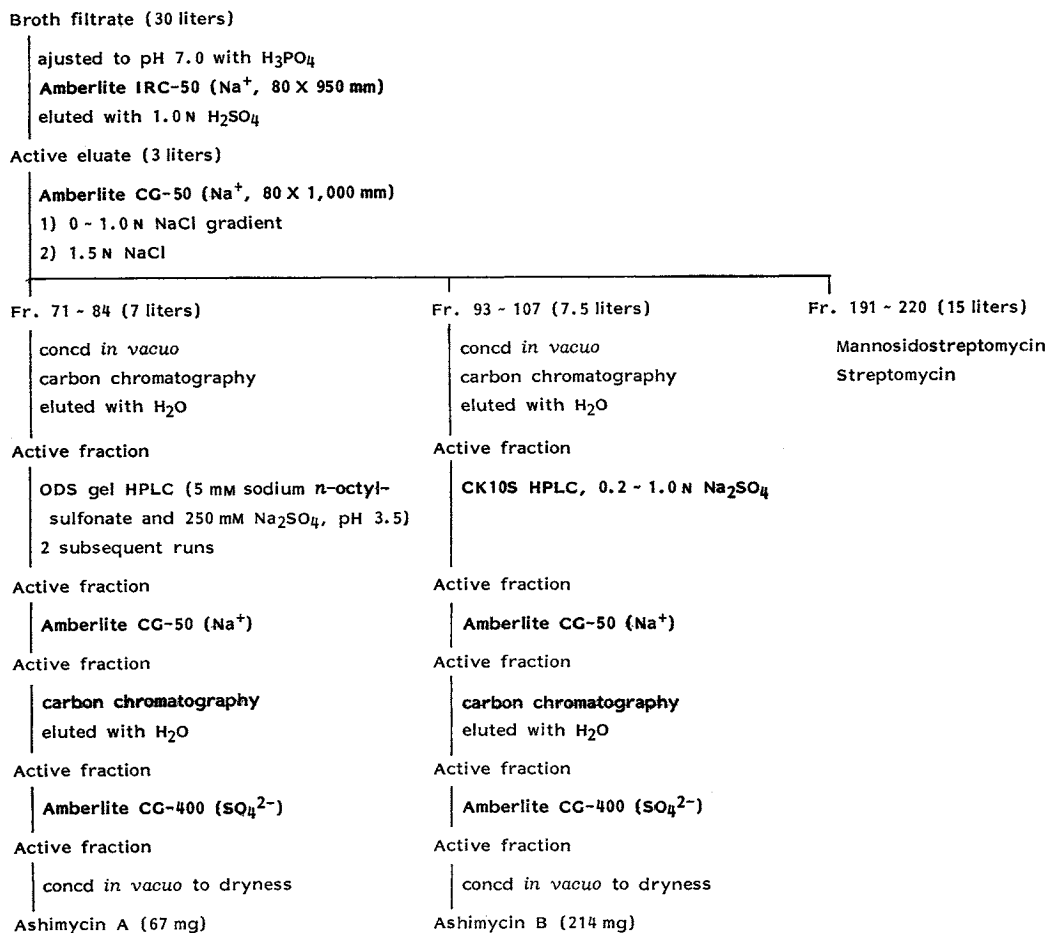
Ashimycins A and B have been isolated from the culture filtrate of *Streptomyces griseus* strain FT3-4 which is being used for commercial production of streptomycin at our company. A well-grown agar slant of the strain FT3-4 was used to inoculate 20 ml of a seed culture medium consisting of corn steep liquor 3.0%, yeast 0.25% and $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ 0.1% (pH 7.5) in a 100-ml Erlenmeyer flask. The flask was shaken on a rotary shaker (220 rpm) at 28°C for 27 hours, and then 4 ml of the culture was transferred to 80 ml of the same medium in a 500-ml Erlenmeyer flask. After shaking at 28°C for 21 hours, 80 ml of the broth was used to inoculate 1,000 ml of the same medium in a 5-liter Erlenmeyer flask, and the fermentation was continued at 28°C for 30 hours. This seed culture was added to a 30-liter jar fermenter containing 20 liters of a production medium composed of soybean meal 5.0%, glucose 5.0%, mannose 0.5% and KH_2PO_4 0.1%, pH being adjusted to 7.6 before sterilization. The final fermentation was carried out at 28°C for 6 days with an agitation rate of 250 rpm and aeration of 8 liters per minute.

Isolation and Purification

The procedures for isolating a crude streptomycin complex and purifying the individual components are schematically shown in Fig. 1. Active substances were traced by the maltol reaction after separation by analytical HPLC (column: CK10S, Mitsubishi Chemical Industries Limited, 7.0×250 mm; solvent: linear gradient from 0.2 to 1.0 M Na_2SO_4 , flow rate: 1.0 ml/minute). The broth filtrate contained the ashimycins in addition to mannosidostreptomycin and streptomycin as shown in Fig. 2.

The broth filtrate was adsorbed on an Amberlite IRC-50 column (Na^+ , 4.8 liters), and elution with

Fig. 1. Isolation scheme for ashimycins A and B.



1.0 N H_2SO_4 gave an active eluate (3 liters) which was concentrated to give a crude powder (269 g). Separation of the individual antibiotics was achieved by Amberlite CG-50 column chromatography (Na^+ , 80 X 1,000 mm) developed with a 0~1.0 N NaCl gradient (72 liters). The eluate was cut into 0.5-liter fractions. Ashimycin A was eluted in fractions 71~84 and ashimycin B in fractions 93~107. Mannosidostreptomycin and streptomycin were eluted by 1.5 N NaCl.

Ashimycin A was further purified by carbon column chromatography (Wako, activated carbon, 40 X 360 mm) eluted with water followed by liquid chromatography on YMC-ODS (Shimadzu, Co. 50 X 500 mm, equilibrated with an aqueous solution containing 5 mM of sodium *n*-octylsulfonate and 250 mM of sodium sulfate adjusted to pH 3.5) developed with the same solution. The purification procedure by liquid chromatography was repeated twice. The concentrated fraction containing ashimycin A was adsorbed on a column of Amberlite CG-50 (Na^+ , 60 X 360 mm) followed by elution with a gradient from 0~1.0 N NaCl. The active fraction was concentrated and was subjected to carbon column chromatography (10 X 64 mm) developed with water. The concentrated active solution of the eluate was adsorbed on a column of Amberlite CG-400 (SO_4^{2-} , 12 X 88 mm) and developed with water to yield a colorless hygroscopic powder of ashimycin A sulfate (67 mg).

Ashimycin B was purified by carbon chromatography (40×360 mm) eluted with water followed by liquid chromatography on CK10S (10 μ m, 50×500 mm) with a 0.2~1.0 N Na₂SO₄ gradient system. The concentrated solution was adsorbed on a column of Amberlite CG-50 (Na⁺, 60×360 mm) and eluted with a gradient from 0~1.0 N NaCl. The active fraction was concentrated and was subjected to carbon column chromatography (10×64 mm) developed with water. The concentrated active solution of the eluate was adsorbed on a column of Amberlite CG-400 (SO₄²⁻, 12×88 mm) and developed with water to yield a colorless powder of ashimycin B sulfate (214 mg).

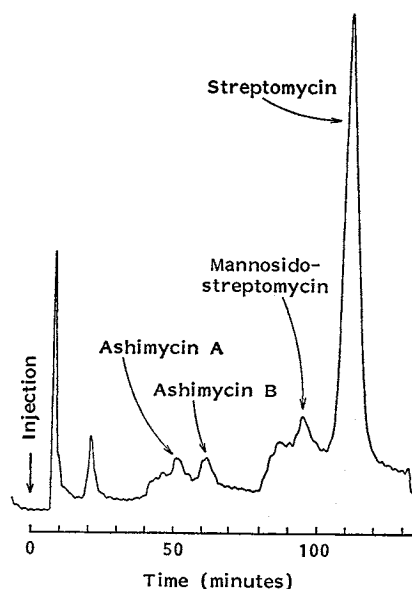
Physico-chemical Properties

Physico-chemical properties of ashimycins A and B are listed in Table 1. Both compounds were soluble in water and insoluble in methanol and chloroform. The molecular formula of ashimycins A and B were determined by secondary ion mass spectrometry (SI-MS) and elemental analysis. The IR absorption at 1660~1630 cm⁻¹ (Figs. 3 and 4) suggested that these antibiotics contained guanidino groups. Ashimycin A was positive to Sakaguchi, maltol and Elson-Morgan reactions, while ashimycin B was negative to Elson-Morgan reaction.

Structures of Ashimycins A and B

In the 400 MHz ¹H NMR spectrum of ashimycin A in D₂O (Fig. 5), three anomeric proton signals were observed at δ 5.30 ($J_{1',2'}=1.4$ Hz, 1'-H), 5.74 ($J_{1'',2''}=3.5$ Hz, 1''-H) and 5.36 (s, 1'''-H). The first two were ascribed to the anomeric protons of streptose and *N*-methylglucosamine, respectively. The last singlet signal at δ_H 5.36 was assigned to the anomeric proton

Fig. 2. HPLC analysis of the fermentation broth of *Streptomyces griseus* FT3-4.



Column: CK10S (7.0×250 mm), mobile phase; 0.2~1.0 M Na₂SO₄, pH 3.5 (gradient), detection; maltol reaction.

Table 1. Physico-chemical properties of ashimycins A and B.

	Ashimycin A	Ashimycin B
Appearance	White amorphous powder	White amorphous powder
MP(°C)	150	205
Optical rotation $[\alpha]_D^{25}$	-37.0° (c 1.0, H ₂ O)	-85.3° (c 0.04, H ₂ O)
SI-MS (m/z, (M+H))	758	640
Molecular formula	C ₂₇ H ₄₇ N ₇ O ₁₈ ·1½H ₂ SO ₄ ·2H ₂ O	C ₂₃ H ₄₁ N ₇ O ₁₄ ·H ₂ SO ₄ ·H ₂ O
Elemental analysis	C H N	C H N
Calcd:	34.47, 5.79, 10.42	36.56, 6.00, 12.98
Found:	34.10, 5.68, 10.15	36.39, 6.04, 13.31
IR ν_{max}^{KBr} cm ⁻¹	3400, 1660, 1630, 1400, 1100, 600	3340, 1660, 1630, 1400, 1100, 600
Color reaction		
Positive:	Sakaguchi, maltol, Elson-Morgan	Sakaguchi, maltol
Negative:		Elson-Morgan

Fig. 3. IR spectrum (KBr) of ashimycin A.

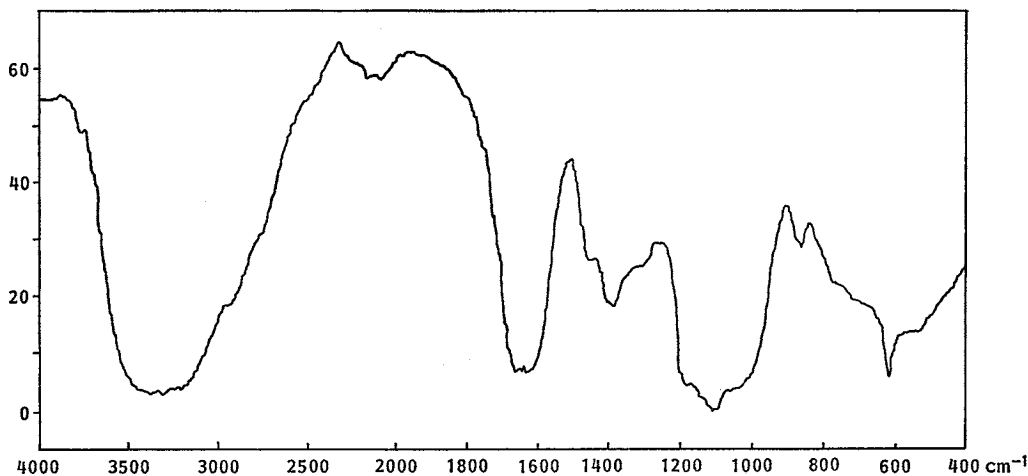
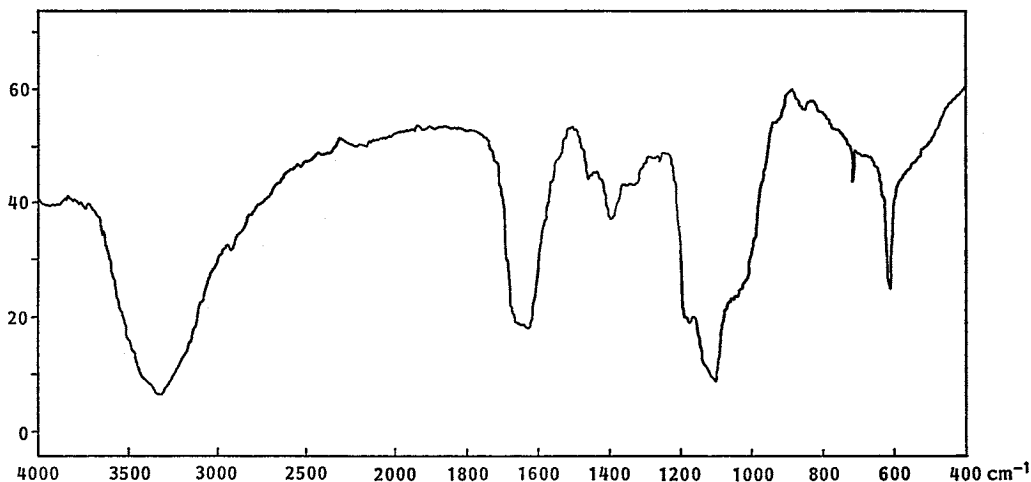
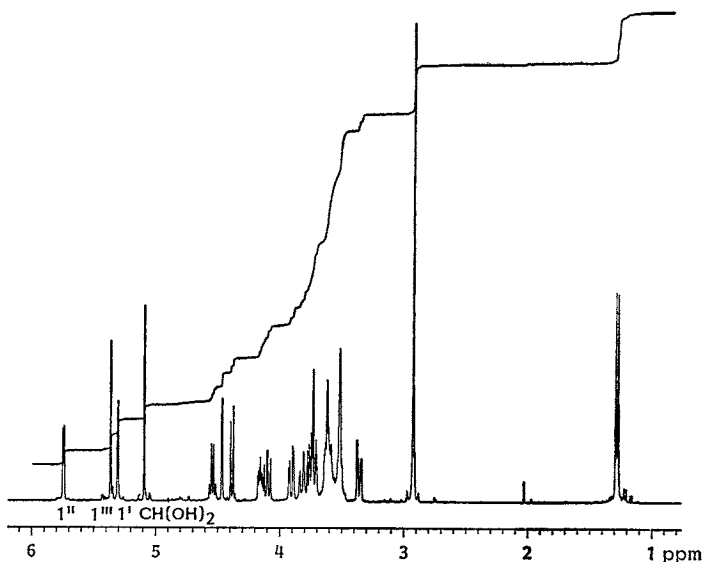


Fig. 4. IR spectrum (KBr) of ashimycin B.



of a new sugar moiety. ^{13}C NMR spectral comparison of ashimycin A and streptomycin (Table 2) suggests that the new antibiotic is structurally very similar to streptomycin except for the presence of a new sugar^{1,2)}. The functionalities of the carbon atoms in the new sugar are as follows; $1 \times$ anomeric carbon, $1 \times \text{COOH}$, $1 \times \text{C-O}$, $2 \times \text{CH-O}$ and $1 \times \text{CH}_2\text{-O}$. The partial structures of the new sugar from C-3''' to C-5''' (Fig. 6) were determined by proton spin decoupling experiments. The ^{13}C chemical shifts of C-4''' (δ 83.3) and C-5''' (δ 62.0) are very similar to those of C-4' and C-5' of 9- β -D-xylofuranosyl adenine (δ 83.0 and 60.0 respectively)³⁾, suggesting that C-4''' must be attached to C-1''' via an ethereal oxygen. The downfield shift of C-4''' is reasonably explained in terms of an alkylation shift. Since C-1''' and C-3''' were not coupled to each other, they must be separated by a quaternary oxycarbon (δ 82.0) and the carboxyl group is therefore linked to C-2'''. Furthermore, the presence of a carboxyl group in ashimycin A was revealed by positive coloration of a hydroxamic acid derivative⁴⁾. The configurations of C-1''' and C-2''' remain to be established.

Fig. 5. 400 MHz ^1H NMR spectrum of ashimycin A.

The position of the new sugar within the molecule was determined as follows. Compared with streptomycin, C-4'' is shifted downfield by 8.5 ppm, and C-3'' and C-5'' are shifted upfield by 1.8 and 1.9 ppm, respectively, in ashimycin A. Thus, the C-4'' position of the streptomycin moiety is glycosylated by the new sugar in ashimycin A. Accordingly, the structure of ashimycin A was determined as shown in Fig. 7. We propose to name the new sugar as ashimose. A sugar with the same planar structure had been prepared by oxidation of methyl β -D-glucopyranoside with oxygen in alkaline, aqueous solution⁵⁾.

In the 200 MHz ^1H NMR spectrum of ashimycin B in D_2O (Fig. 8), two anomeric proton signals were observed at δ 5.35 ($J_{1',2'}=1.5$ Hz, 1'-H) and 5.70 ($J_{1'',2''}=3.4$ Hz, 1''-H). The ^{13}C NMR spectrum (Table 2) suggests that ashimycin B is structurally very similar to streptomycin except for the presence of two new signals due to carbonyl (δ_{C} 171.5) and methylene (δ_{C} 58.9) carbons, assignable to a glycolic acid moiety (δ_{C} C-1, 177.0 and C-2, 60.5)⁶⁾. The downfield shifts of NCH_3 by 8.2 ppm and of C-2'' (δ_{C} 66.4)

Table 2. ^{13}C NMR shifts of ashimycins A and B^a.

Carbon	Ashimycin A	Ashimycin B	Streptomycin
1	59.6	59.7	59.7
2	72.1	71.7	71.5
3	59.3	59.3	59.0
4	76.6	78.7	78.9
5	74.3	74.0	74.2
6	72.3	72.4	72.4
C=NH (1)	159.1	159.1	159.1
C=NH (3)	158.3	158.5	158.6
1'	107.3	106.4	106.7
2'	81.6	84.1	85.3
3'	83.4	83.2	83.1
4'	78.8	78.0	78.3
5'	12.3	13.7	13.4
CHO (3')	89.9	90.5	90.5
1''	92.6	95.9	95.2
2''	62.0	66.4	62.3
3''	68.6	68.4	70.4
4''	78.6	71.0	70.1
5''	71.8	73.5	73.7
6''	60.2	61.3	61.2
NCH_3 (2'')	32.7	41.3	33.1
C=O		171.5	
CH_2OH		58.9	
1'''	106.7		
2'''	82.0		
3'''	74.3		
4'''	83.3		
5'''	62.0		
COOH	177.2		

From internal dioxane in ppm. pD 5.5.

^a For the assignments of streptomycin and related substances, see refs 1 and 2.

Fig. 6. Partial structures of ashimycin A.

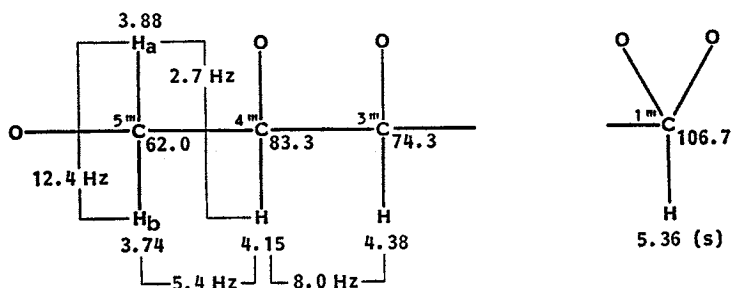
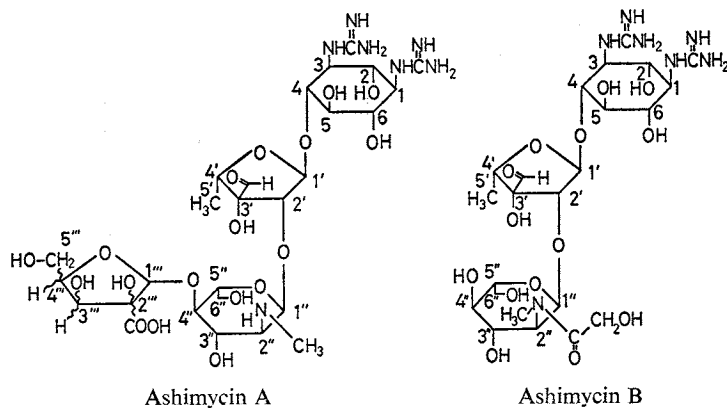
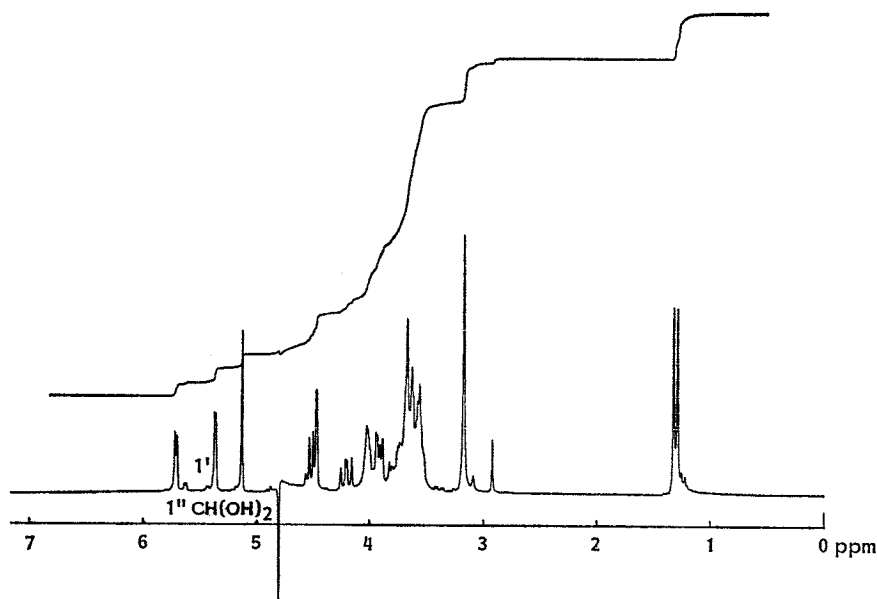


Fig. 7. Structures of ashimycins A and B.

Fig. 8. 200 MHz ¹H NMR spectrum of ashimycin B.

by 4.1 ppm in the ¹³C NMR data of ashimycin B suggest that the *N*-methyl group is acylated with glycolic acid in ashimycin B. In agreement with this conclusion, long range selective proton decoupling (LSPD) of the *N*-methyl proton collapsed the carbonyl carbon signal (δ_c 171.5) to a broad

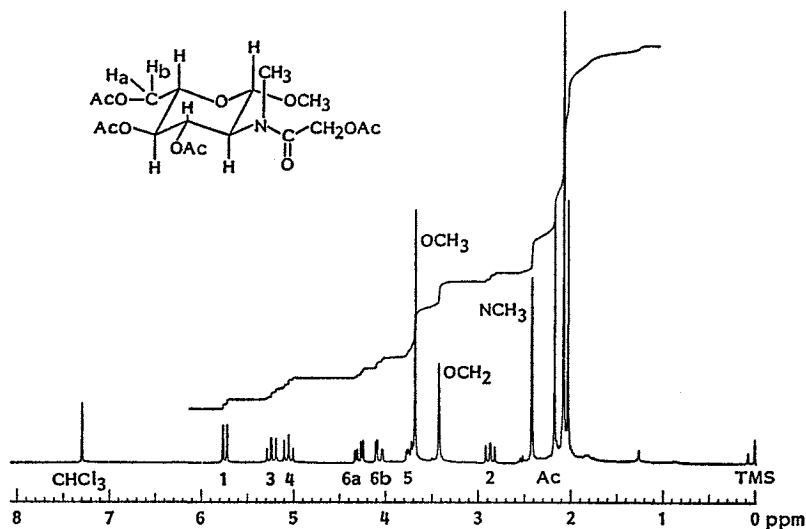
Fig. 9. 200 MHz ^1H NMR spectrum of methyl *N*-glycolyl-*N*-methylglucosaminide peracetate.

Table 3. Antibacterial activities of ashimycins A and B.

Test organisms	Ashimycin A	Ashimycin B	Streptomycin
<i>Staphylococcus aureus</i> Smith I	50	12.5	1.56
<i>S. aureus</i> FDA 209P	25	25	1.56
<i>S. aureus</i> Apo-1	50	50	3.13
<i>S. epidermidis</i> 109	50	100	100
<i>Micrococcus flavus</i> FDA 16	100	25	1.56
<i>Bacillus anthracis</i>	50	12.5	0.78
<i>B. subtilis</i> PCI 219	25	6.25	0.39
<i>Corynebacterium bovis</i> 1810	100	25	1.56
<i>Escherichia coli</i> K-12 R-5	50	100	100
<i>E. coli</i> NIHJ JC-2	50	25	3.13
<i>Klebsiella pneumoniae</i>	200	25	1.56
<i>Proteus vulgaris</i> OX19	50	12.5	1.56
<i>P. rettgeri</i> GN311	25	12.5	0.78

MIC ($\mu\text{g/ml}$) in Mueller-Hinton agar.

doublet signal ($J=4.9$ Hz). Furthermore, the *N*-methyl proton (δ_{H} 3.16) of ashimycin B is shifted downfield by 0.26 ppm as compared with the corresponding proton of streptomycin (δ_{H} 2.90).

In order to confirm the structure of ashimycin B, it was subjected to acid catalyzed hydrolysis. Ashimycin B was dissolved in 1 N H_2SO_4 and allowed to stand at room temperature for 18 hours. After removal of precipitated streptidine sulfate by filtration, the filtrate was treated with concentrated HCl for 2 hours at 100°C. The residue was passed through a column of Dowex 50 (H^+) and then an *N*-methylglucosamine derivative was eluted with 4 N NH_4OH . Treatment of the dried hydrolysate with HCl-methanol solution, followed by acetylation with acetic anhydride in the presence of anhydrous sodium acetate gave a crude powder of methyl *N*-glycolyl-*N*-methylglucosaminide peracetate. This material was purified by silica gel chromatography eluting with CHCl_3 - CH_3OH (100:1). Physico-chemical properties of the peracetate were as follows: $[\alpha]_{\text{D}}^{25} -22.8^\circ$ (c 0.07, CHCl_3); SI-MS m/z 434 ($\text{M}+\text{H}$) $^+$; the 200 MHz ^1H NMR spectrum in CDCl_3 is shown in Fig. 9.

Anal Calcd for $C_{18}H_{27}NO_{11} \cdot H_2O$: C 47.89, H 6.47, N 3.10.

Found: C 48.36, H 6.11, N 3.03.

Thus the glycol unit is attached to NCH_3 by amide bond. Consequently the structure of ashimycin B was established as shown in Fig. 7.

Antibacterial Properties

The MICs of ashimycins A and B assayed by the agar dilution method are given in Table 3. They showed similar activities against *Staphylococcus aureus* Apo-1. Ashimycin A showed higher activities than ashimycin B except for *Staphylococcus epidermidis* 109 and *Escherichia coli* K-12 R-5.

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